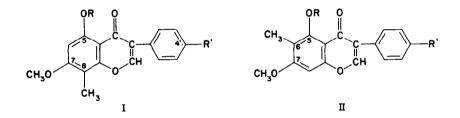
Tetrahedron Letters No. 20, pp. 889-891, 1962. Pergamon Press Ltd. Printed in Great Britain.

RING ISOMERIZATION OF C-METHYL ISOFLAVONES NEW SYNTHESES OF 2-HYDROXY ISOFLAVANONES Lorand Farkas, Jozsef Varady and Agnes Gottsegen Institute of Organic Chemistry, Technical University, Budapest (Received 13 July 1962)

ACCORDING to our present knowledge,¹ the rearrangement of 5-hydroxy-7,4'dimethoxy-8-methyl isoflavone catalysed by acid gives a mixture of 6- and 8-methyl-5,7,4'-trihydroxy isoflavones.

Our recent investigations proved, however, that this rearrangement takes place unequivocally in the presence of a basic catalyst.



Thus, on the effect of potassium ethylate, 5-hydroxy-7-methoxy-8-methyl isoflavone (I; R = R' = H, m.p. 169-170[°]) converts into 5-hydroxy-7-methoxy-6-methyl isoflavone (II; R = R' = H, m.p. 176-177[°], lit. m.p.² 170-172).

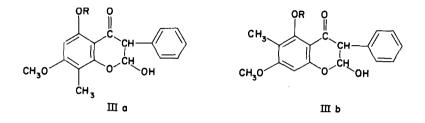
On reacting, in turn, 5,7-dimethoxy-8-methyl isoflavone² (I; $R = CH_3$, R' = H) with potassium ethylate; ring isomerization can not take place as

¹ W.B. Whalley, <u>J. Chem. Soc.</u> 3366 (1953).

² M.O. Farooq, W. Rahman and KH.T. Nasim, <u>J. Org. Chem.</u> <u>27</u>, 944 (1962).

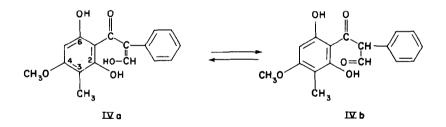
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the hydroxyl group on C_5 is not free, and thus, after careful neutralization of the reaction mixture, 2-hydroxy-5,7-dimethoxy-8-methyl isoflavanone (IIIa; R = CH₃, m.p. 191-192[°]) was obtained.



The same compound was prepared also from 2-hydroxy-4,6-dimethoxy-3methyl-desoxybenzoin,³ subjecting it to a Claisen condensation (ethyl formate, metallic sodium). Compounds IIIa ($R = CH_3$) prepared by both methods proved to be completely identical.

Thus, we presume that on the effect of potassium ethylate, the isoflavone ring splits also on the case of compound I (R = R' = H), and that, on cautiously neutralizing the produced α -hydroxymethylene ketone (IVa) and β -ketone aldehyde (IVb), respectively, they form, owing to free rotation,



an isoflavanone ring (IIIb; R = H) with the hydroxyl group of C₆ of higher reactivity. On stronger acidification, however, this product converts,

under loss of water, into isoflavone II (R = R' = H).

³ R. Iengar, A.C. Mehta, T.R. Seshadri and S. Varadarajan, <u>J. Sci. Ind. Res.</u> (India) 13B, 166 (1954).

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Similary, we succeeded in unequivocally isomerizing 5-hydroxy-7,4'dimethoxy-8-methyl isoflavone (I; R = H, $R' = 0CH_3$, m.p. 169-170°, lit. m.p.⁴ 164-166°) to 5-hydroxy-7,4'-dimethoxy-6-methyl isoflavone (II; R = H, $R' = 0CH_3$, m.p. 204-205°, lit. m.p.⁵ 200-201°) by treatment with potassium ethylate.

Our investigations in order to extend this reaction are in progress. Our communication will shortly be published in detail in <u>Chemische Berichte</u>.

⁴ T.R. Seshadri and S. Varadarajan, <u>Proc. Ind. Acad. Sci.</u> <u>37A</u>, 145 (1953).

⁵ A.C. Mehta and T.R. Seshadri, <u>J. Chem. Soc.</u> 3823 (1954).